BALTIC SEA ACIDIFICATION BY SO\textsubscript{X} AND NO\textsubscript{X} FROM LAND AND SHIPPING: MODELLING OF PAST AND POTENTIAL FUTURE SCENARIOS

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Outline

• SO$_X$ and NO$_X$:
  • Shipping emissions and regulations
  • Scrubbers for SO$_X$ reduction
  • Terrestrial sources
• Baltic Sea modelling
  • Hindcast
  • Future scenarios
CO₂ is not the only acidic gas generated by ship motors

While the emitted CO₂ is distributed globally, much of the SOₓ and NOₓ is deposited within a few hundred km of the source. This is also true of other plume components.
Emissions of $SO_X$ are subject to increasing regulation, particularly in Sulphur Emission Control Areas (SECA)

SECA development is focused on heavily trafficked coastal areas. Controls on $NO_X$ emissions, applying to new builds, are being introduced more slowly.
The new SECA regulations with a maximum 0.1% sulphur in marine fuels, which came into effect in January 2015, have been extremely controversial in Northern Europe:

- Changing from 1% sulphur to 0.1% doubles the fuel cost
- The limited scope of SECA risks distorting competition (e.g. SECA in Northern Europe but not the Mediterranean)

The shipping industry is responding in two ways:

- Investment in new fuels such as Liquefied Natural Gas and methanol (mostly for new builds)
- Investment in scrubbers to remove SOx from the smokestack gases
Shipping in SECA 2015: buy expensive fuel, or install a scrubber?

(a) fuel < 0.1% S → low-sulphur emissions

Fuel costs double on switching from 1% S to 0.1% S

(b) fuel 2-3% S → exhaust gas → scrubber → low-sulphur emissions

The scrubber absorbs $SO_x$ in a fine spray of seawater

“scrubber water”
- acidic (pH 3)
- contains metals and other toxins

What happens to this??
SO\textsubscript{x} in the exhaust gases is absorbed by a counterflow of seawater:

\[ \text{SO}_2 + \text{H}_2\text{O} = \text{H}_2\text{SO}_3 = \text{H}^+ + \text{HSO}_3^- \quad \text{SO}_3 + \text{H}_2\text{O} = \text{H}_2\text{SO}_4 = 2 \text{H}^+ + \text{SO}_4^{2-} \]

This results in water with pH \( \approx 3 \)

Open loop: acidified water is discharged

Closed loop: acidified water is neutralised
The development of marine $SO_X$ regulations is very slow in comparison with land transport.

Note the log scale!

Progress in regulating marine $NO_X$ emissions is even slower.
Sources of $SO_\text{x}$ and $NO_\text{x}$ in the Baltic Sea

Reconstructed annual atmospheric deposition of $SO_\text{x}$ and $NO_\text{x}$ from land and sea within ± one standard deviation for the three model sub-basins Kattegat (ka), Eastern Gotland Basin (go) and Bothnian Bay (bb) (Omstedt et al., 2015)
The PROBE model, developed by Omstedt and colleagues, divides the waters from the Kattegat to the Bothnian Bay into 13 basins. The model has been optimised with the help of historical data.

Atmospheric modelling uses terrestrial and marine emission data, together with a chemical transport model to calculate deposition of $SO_X$ and $NO_X$.

Our modelling results are presented as differences from a "base case" where there is no deposition of $SO_X$ or $NO_X$. 
Hindcast: shipping’s contribution to alkalinity reduction

The accumulated sink/source of total alkalinity during the 1860-2013 period based on atmospheric deposition of NH$_x$, NO$_x$/SO$_x$ total emissions and from shipping only (Omstedt et al., 2015)
Modelling of future scenarios

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<th>Scenario no.</th>
<th>Shipping not using wet scrubbers</th>
<th>Shipping using wet scrubbers</th>
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- Climate forcing follows RCP 4.5
- Traffic increases by 2.5% per year (cargo). 3.9% per year (passengers)
- No additional NOX regulation is assumed
The proportion of strong acid deposition to the Baltic Sea due to shipping in the five future scenarios (Turner et al., in preparation)
Modelled future changes in pH and alkalinity due to shipping in surface waters of the Arkona Basin, East Gotland Basin and the Bothnian Bay, according to the five future scenarios (Turner et al., in preparation)
Modelled future changes in pH and alkalinity in surface waters of the Arkona Basin, East Gotland Basin and the Bothnian Bay, assuming that all ships use open-loop scrubbers (scenario 5) (Turner et al., in preparation).
Future perspectives

• Acidification is probably not the major environmental consequence of scrubber operation

• In addition to the low pH, publicly available analyses from early scrubber systems show that the outlet water contained high concentrations of metals such as copper and zinc, as well as toxic organic compounds

• Research is needed on the environmental consequences of these releases, particularly in heavily trafficked areas

• The results of such research should be used to develop improved regulation of scrubber use